



Tetrahedron Letters 46 (2005) 6705-6708

Tetrahedron Letters

Palladium-catalyzed coupling reaction of propargylic oxiranes with arylboronic acids in aqueous media

Masahiro Yoshida, a,* Hirofumi Uedab and Masataka Iharab,*

^aGraduate School of Pharmaceutical Sciences, The University of Tokushima, 1-78-1 Sho-machi, Tokushima 770-8505, Japan ^bGraduate School of Pharmaceutical Sciences, Tohoku University, Aobayama, Sendai 980-8578, Japan

> Received 27 June 2005; revised 25 July 2005; accepted 26 July 2005 Available online 10 August 2005

Abstract—A novel type of coupling reaction has been developed by the palladium-catalyzed reaction of propargylic oxiranes with arylboronic acids, in which *anti*-substituted 4-aryl-2,3-allenols were produced in a highly diastereoselective manner. A chiral-substituted allene has been synthesized from the reaction of a chiral propargylic oxirane without loss of the chirality. © 2005 Elsevier Ltd. All rights reserved.

The functionalized allenes are versatile building blocks for organic synthesis because of the inherent reactivity for their axially chiral backbones. Furthermore, a large number of natural products containing an allene moiety have been isolated, and most of these have axial chirality.2 Extensive studies for the synthesis of allenes have been undertaken from these reasons, and S_N2'-type reactions of propargylic oxiranes are one of the most successful procedures to afford 2,3-allenols.³ Propargylic oxiranes generally reacted with organocopper,4 magnesium⁵ and metal hydride⁶ to produce the substituted 2,3-allenols in moderate to high diastereoselectivity. It is also known that palladium complexes catalyze the reactions of propargylic oxiranes with organozine,⁷ stannane⁸ and carbon monooxide⁹ yielding the corresponding 2,3-allenols. These reactions are normally required to be carried out under anhydrous conditions, and some of the organometallic reagents are unstable, expensive and toxic. Organoboronic acids are widely used reagents in organic synthesis because of their stability, commercial availability and nontoxicity. However, to the best of our knowledge, there are no examples about the reaction of propargylic oxiranes with organoboronic acids.

Herein, we describe a novel type of coupling reaction of propargylic oxiranes with arylboronic acids by palladium catalyst. The reactions can be carried out in aqueous media to produce the aryl-substituted 2,3-allenols with high *anti*-diastereoselectivity.

The initial reactions were carried out using phenylsubstituted propargylic oxirane $1a^{10}$ and 2-methylphenylboronic acid (2a). When a mixture of 1a and 2a were subjected to the reaction with 10 mol%Pd(PPh₃)₄ in dioxane at 80 °C for 2 h, the aryl-substituted 2,3-allenol 3aa having anti-geometry was obtained in 81% yield (Table 1, entry 1). The stereochemistry of 3aa was determined unambiguously by NOESY correlation of dihydrofuran 4aa, which was produced from the reaction of 3aa with AgNO₃ (Scheme 1). The similar result was obtained by using Pd₂(dba)₃·CHCl₃ with P(o-Tol)₃ (entry 2), but poor results were obtained in the presence of bidentate phosphine ligands (entries 3– 5). It was found that the yields of 3aa were increased in the presence of water (entries 6–8), and the product was produced in 92% yield when the reaction was employed in dioxane-H₂O (2:1) (entry 7). Since the obtained 2,3-allenol had all anti-geometry in any conditions, it was ascertained that this addition reaction proceeds in a high diastereoselective manner.

A series of substituted arylboronic acids **2b**—**f** were then subjected to the reaction. The corresponding products **3ab** and **ac** were obtained in good yields by the reactions of **1a** with 2- and 4-methoxyphenylboronic acids (**2b** and **c**) (Table 2, entries 1 and 2). 1-Naphthalene- and phenylboronic acids (**2d** and **e**) were also efficiently transformed to the coupled products **3ad** and **ae** (entries

Keywords: Palladium; Addition; Oxiranes; Boronic acids; Propargylic compounds.

^{*}Corresponding authors. Fax: +81 88 6337294; e-mail: yoshida@ph.tokushima-u.ac.jp

Table 1. Palladium-catalyzed coupling of propargylic oxirane **1a** with 2-methylphenylboronic acid **(2a)**

0 1a		nol % Pd(0) vent, 80 °C	Ph
Entry	Palladium catalyst	Solvent	Yield
			(%)
1	$Pd(PPh_3)_4$	Dioxane	81
2^{a}	Pd ₂ (dba) ₃ ·CHCl ₃ , P(o-Tol) ₃	Dioxane	73
3 ^b	Pd ₂ (dba) ₃ ·CHCl ₃ , dppe	Dioxane	16
4 ^b	Pd ₂ (dba) ₃ ·CHCl ₃ , dppp	Dioxane	38
5 ^b	Pd ₂ (dba) ₃ ·CHCl ₃ , dppf	Dioxane	42
6	$Pd(PPh_3)_4$	$Dioxane-H_2O = 9:1$	91
7	$Pd(PPh_3)_4$	Dioxane $-H_2O = 2:1$	92
8	Pd(PPh ₃) ₄	Dioxane– $H_2O = 1:2$	61

^a 5 mol% palladium and 40 mol% ligand were used.

Scheme 1.

Table 2. Palladium-catalyzed coupling of propargylic oxirane 1a with arylboronic acids $2b\!-\!f$

Entry	Boronic acid	Product ^a	Yield (%)
1	2-Methoxyphenylboronic acid (2b)	$3ab^a$	77
2	4-Methoxyphenylboronic acid (2c)	3ac ^a	81
3	1-Naphthaleneboronic acid (2d)	$3ad^a$	86
4	Phenylboronic acid (2e)	3ae ^a	55
5	3-Nitrophenylboronic acid (2f)	3af ^b	38

^a The stereochemistry of each product was tentatively assigned by comparison of its NMR spectra with **3aa** and **af**.

3 and 4). When 3-nitrophenylboronic acid (2f) having an electron-withdrawing group was subjected to the reaction, the yield of the resulting product 3af was slightly lowered (entry 5).

Results of reactions of various propargylic oxiranes 1b-g with 2-methylphenylboronic acid (2a) are summarized in Table 3. When the reactions of substrates 1b and c

Table 3. Reactions of various propargylic oxiranes **1b–g** with 2-methylphenylboronic acid (**2a**)^a

Entry	Substrate		Product ^b		Yield (%)
1	0,	_Ph 1b	OH Ar	h Bba ^c	64
2		Ph 1c		Ph Bca ^c	83
3		_Bu	OH Ar B	u 3da ^c	77
4		OMe 1e	OH Ar	_OMe Bea ^c	85
5		TMS	OH Ar	Bfa ^d	43
6	H, Ph H	Ph 1g	OH Ar	Ph 3ga ^c	50

^a Reactions were carried not with 2-methylphenylboronic acid (**2a**) in the presence of 10 mol% Pd(PPh₃)₄ in dioxane–H₂O (2:1) at 80 °C.

^c The stereochemistry of each product was tentatively assigned by comparison of its NMR spectra with **3aa**, **af** and **fa**.

possessing a five- and a seven-membered ring were carried out, the coupled 2,3-allenols **3ba** and **ca** were obtained in 64% and 83% yields, respectively (entries 1 and 2). Substrates **1d** and **e**, having a butyl and a methoxymethyl group at the terminal position, were uneventfully reacted with **2a** to afford the corresponding products **3da** and **3ea** in good yields (entries 3 and 4). When TMS-substituted propargylic oxirane **1f** was subjected to the reaction, the desilylated 2,3-allenol **3fa** was predominantly produced (entry 5). The reaction of acyclic substrate **1g** also uneventfully proceeded to yield the corresponding coupled product **3ga** in moderate yield (entry 6).

We further attempted the reaction of enantiomerically enriched propargylic oxirane (1R,2R)-1a (Scheme 2). When (1R,2R)-1a¹² (92% ee) was reacted with 2-methylphenylboronic acid (2a), the corresponding chiral coupled product (1R,2R)-3aa¹³ was provided in 92% yield. The enantiomeric excess of (1R,2R)-3aa was determined as 92%, and the result showed that the reaction proceeded with complete transferring chirality.

^b 5 mol% palladium and 20 mol% ligand were used.

^b The stereochemistry was determined unambiguously by NOESY correlation of dihydrofuran **4af**, which was produced from the reaction of **3af** with AgNO₃ and CaCO₃.

 $^{^{\}rm b}$ Ar = 2-methylphenyl.

^d The stereochemistry was determined unambiguously by NOESY correlation of dihydrofuran **4fa**, which was produced from the reaction of **3fa** with AgNO₃ and CaCO₃.

Scheme 2.

Scheme 3.

Plausible mechanism for the formation of aryl-substituted 2,3-allenols 3 is shown in Scheme 3. In the first step, regio- and stereoselective *anti*- S_N2' attack of palladium catalyst¹⁴ on the propargylic oxirane 1 takes place to yield the twitter ionic allenylpalladium species 5, which was further hydrated in the presence of H_2O to form the allenylpalladium hydroxide $6.^{15}$ Transmetallation of 6 with arylborate $7,^{16}$ derived from arylboronic acid 2 and H_2O , and then reductive elimination of palladium from the resulting intermediate 8 diastereoselectively produces *anti*-coupled 4-aryl-2,3-allenol 3.

In conclusion, the effort described above has led to the discovery of a palladium-catalyzed coupling reaction occurring between propargylic oxiranes and arylboronic acids. The process can be carried out in aqueous conditions to yield *anti*-substituted 4-aryl-2,3-allenols in a highly diastereoselective manner. Furthermore, the chiral-substituted allene has been synthesized from the chiral propargylic oxirane without loss of the chirality. Continuing studies probing the scope and synthetic applications of this reaction are now in progress.

Acknowledgement

This study was supported in part by a Grant-in-Aid for the Encouragement for Young Scientists (B) from the Japan Society for the Promotion of Science (JSPS) (for M.Y.).

References and notes

- For selected reviews, see: (a) Marshall, J. A. Chem. Rev. 2000, 100, 3163; (b) Zimmer, R.; Dinesh, C. U.; Nandanan, E.; Khan, F. A. Chem. Rev. 2000, 100, 3067; (c) Ma, S. Acc. Chem. Res. 2003, 36, 701; (d) Hashmi, A. S. K. Angew. Chem., Int. Ed. 2000, 39, 3590; (e) Hoffmann-Röder, A.; Krause, N. Angew. Chem., Int. Ed. 2002, 41, 2933; (f) Krause, N.; Hoffmann-Röder, A.; Canisius, J. Synthesis 2002, 1759; (g) Hoffmann-Röder, A.; Krause, N. Angew. Chem., Int. Ed. 2004, 43, 1196.
- (a) Krause, N.; Hoffmann-Röder, A. Allenic Natural Products and Pharmaceuticals. In Modern Allenic Chemistry; Krause, N., Hashmi, A. S. K., Eds.; Wiley-VCH: Weinheim, 2004, p 997; (b) Landor, S. R. Naturally Occurring Allenes. In The Chemistry of the Allenes; Landor, S. R., Ed.; Academic Press: London, 1982, p 679; (c) Claesson, A. Biologically Active Allenes. In The Chemistry of the Allenes; Landor, S. R., Ed.; Academic Press: London, 1982, p 709; (d) Robinson, C. H.; Covey, D. F. Biological Formation and Reactions. In The Chemistry of Ketenes, Allenes and Related Compounds; Patai, S., Ed.; Wiley: Chichester, 1980, p 451.
- Alexakis, A.; Marek, I.; Mangeney, P.; Normant, J. F. Tetrahedron 1991, 47, 1677.
- (a) Vermeer, P.; Meijer, J.; De Graaf, C.; Schreures, H. Rec. Trav. Chim. Pays-Bas 1974, 93, 46; (b) Oehlschlager, A. C.; Czyzewska, E. Tetrahedron Lett. 1983, 24, 5587; (c) Johnson, C. R.; Dhanoa, D. S. J. Org. Chem. 1987, 52, 1885; (d) Marshall, J. A.; Pinney, K. G. J. Org. Chem. 1993, 58, 1885; (e) Bertozzi, F.; Crotti, P.; Macchia, F.; Pineschi, M.; Arnold, A.; Feringa, B. L. Tetrahedron Lett. 1999, 40, 4893.
- Fürstner, A.; Méndez, M. Angew. Chem., Int. Ed. 2003, 42, 5355.
- (a) Mori, K. Tetrahedron 1974, 30, 1065; (b) Ito, M.; Hirata, Y.; Tsukida, K.; Tanaka, N.; Hamada, K.; Hino, R.; Fujiwara, T. Chem. Pharm. Bull. 1988, 36, 3328; (c) Braumeler, A.; Brade, W.; Haag, A.; Eugster, C. H. Helv. Chim. Acta 1990, 73, 700.
- Kleijn, H.; Meijer, J.; Overbeek, G. C.; Vermeer, P. Rec. Trav. Chim. Pays-Bas 1982, 101, 97.
- Kjellgren, J.; Sundén, H.; Szabó, K. J. J. Am. Chem. Soc. 2005, 127, 1787.
- 9. Knight, J. G.; Ainge, S. W.; Baxter, C. A.; Eastman, T. P.; Harwood, S. J. J. Chem. Soc., Perkin Trans. 1 2000, 3788.
- 10. Substrate 1 was easily prepared by the epoxidation of the corresponding known energy with *m*-CPBA in moderate to good yields.
- 11. Typical procedure: to a stirred solution of propargylic oxirane 1a (45.0 mg, 0.227 mmol) in 1,4-dioxane (2.0 ml) and H₂O (1.0 ml) were added 2-methylphenylboronic acid (2a) (92.8 mg, 0.682 mmol) and $Pd(PPh_3)_4$ (26.3 mg,22.7 µmol) at rt, and the stirring was continued for 1.5 h at 80 °C. After filtration of the reaction mixture using small amount of silica gel, the mass was extracted with AcOEt. The combined filtrates were washed with 10% aqueous NaOH and brine, and the residue upon workup was chromatographed on silica gel with hexane-AcOEt (90:10, v/v) as eluent to give the 4-aryl-2,3-allenol 3aa (60.3 mg, 92%) as colourless crystals; mp 98-99 °C; IR (neat) 3342, 2933, 1948, 1595 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.30–7.15 (9H, m), 4.12–4.05 (1H, m), 2.62–2.55 (1H, m), 2.20 (3H, s), 2.24-2.05 (2H, m), 1.95-1.85 (1H, m), 1.92 (1H, s), 1.85–1.75 (1H, m), 1.62–1.38 (3H, m); NMR (100 MHz, CDCl₃) δ 195.4, 137.3, 136.5, 136.5, 130.3, 130.2, 128.3, 127.5, 126.7, 126.5, 125.8, 111.5, 110.9, 69.6, 36.6, 30.1, 27.1, 23.9, 20.2; MS m/z 290 (M⁺); HRMS m/z calcd for C₂₁H₂₂O: 290.1671 (M⁺). Found: 290.1688.

- 12. Chiral substrate (1*R*,2*R*)-1a was prepared by the asymmetric epoxidation of the corresponding energy using Shi's fructose-derived chiral ketone with oxone; Wang, Z.-X.; Cao, G.-A.; Shi, Y. *J. Org. Chem.* 1999, 64, 7646.
- 13. Enantiomeric excess of the obtained (1*R*,2*R*)-3aa was determined by HPLC analysis (DAICEL CHIRALCEL AD).
- 14. It is known that chiral allenes can be synthesized from the palladium-catalyzed reaction of chiral propargylic compounds via stereoselective S_N2' attack of palladium catalyst: (a) Elsevier, C. J.; Stehouwer, P. M.; Westmijze,
- H.; Vermeer, P. J. Org. Chem. 1983, 48, 1103; (b) Marshall, J. A.; Adams, N. D. J. Org. Chem. 1997, 62, 367; (c) Dixneuf, P.; Guyot, T.; Ness, M. D.; Roberts, S. M. Chem. Commun. 1997, 2083; (d) Konno, T.; Tanikawa, M.; Ishihara, T.; Yamanaka, H. Chem. Lett. 2000, 1360.
- 15. From the results that the reaction smoothly proceeded in the absence of water (Table 1, entries 1–5), the direct transmetallation of cationic allenylpalladium 5 to intermediate 8 would also be possible.
- 16. Kotha, S.; Lahri, K.; Kashinath, D. K. *Tetrahedron* **2002**, 58, 9633